Mechanistic insight into the cinnamylmetal-thioacetal reaction employing 2-acetoxy-2-phenylacetaldehyde monothioacetal

Tsuneo Sato and Junzo Otera

Department of Applied Chemistry, Okayama University of Science, Ridai-cho, Okayama 700 (Japan) (Received December 1, 1993; in revised form January 12, 1994)

Abstract

The reaction of 2-acetoxy-2-phenylacetaldehyde monothioacetal with cinnamylsilane and -tin serves as a model system for disclosing the simple diastereoselectivity of unambiguously assigned S_N 1- and S_N 2-1ike reactions between thioacetal and prochiral allylmetals.

Key words: Tin; Silicon; Cinnamylmetal-thioacetal reaction; 2-Acetoxy-2-phenylacetaldehyde monothioacetal

1. Introduction

The last two decades have witnessed the utility of Lewis acid-promoted nucleophilic substitution of acetals in organic synthesis [1]. The stereochemistry of these reactions has received special attention in view of "acyclic stereoselection" and thus elucidation of the reaction mechanism is of central interest today [2]. In general, the reaction consists of coordination of a Lewis acid to an acetal inducing activation of this substrate and subsequent attack by a nucleophile either through an S_N 1-1ike or S_N 2-1ike pathway. The coordination step was studied with cyclic acetals by NMR spectroscopy by Denmark et al. [3]. Some information is also available on the $S_N 1/S_N 2$ alternative problem. Noyori et al. [4] put forth the S_N1 pathway for the trimethylsilyl triflate-promoted reaction. On the other hand, it has generally been recognized that both mechanisms are feasible depending on the conditions [2,5]. Moreover, formation of the S_N 1-1ike solvent separated ion pair or the S_N2-1ike contact ion-pair has also been suggested [2,5d]. Despite such considerable advances, we have little knowledge about the reaction course beyond this stage which is crucial for elucidating the stereochemistry [6]. This is ascribed to a failure to create reaction systems whose mechanisms are unambiguously assigned.

Reactions employing prochiral silyl [1,7] and tin [8] nucleophiles are particularly important because of their synthetic utility. However, as pointed out by Heathcock et al. [5d], mechanistic studies with the acetals suffer from a severe drawback since the diastereotropic alkoxy groups are not easily differentiated especially in an $S_N 2$ reaction. According to the more recent statement by Sammakia et al. [2b], "it is difficult to draw any firm conclusions about the mechanism of related acetals". α -Chiral acetals are promising for probing this mechanistic dichotomy. We disclosed earlier that the reaction of 2-acetoxy-2-phenylacetaldehyde monothioacetal (1) with allyltins proceeds through an S_N 2-1ike pathway while an S_N 1-1ike mechanism operates in the reaction with allylsilanes [9]. Accordingly, we have been interested in using 1 to elucidate the stereochemistries resulting from unambiguously assigned S_N1-1ike and S_N 2-1ike reactions.

Another requirement to be satisfied in order to reach reliable conclusions is to employ stereochemically rigid allylmetals. Since crotyltins are very labile to double bond isomerization in the presence of a Lewis acid [10], we chose stereochemically rigid (E)-cinnamyltributyltin (2) and -trimethylsilane (3) [11^{*}].

Correspondence to: Dr. J. Otera.

^{*} Reference number with asterisk indicates a note in the list of references.







Scheme 3.

56

2. Results and discussion

2.1. Reaction of 2-acetoxy-2-phenylacetaldehyde monothioacetal (1) with allylmetals (2,3)

Treatment of the u- and l-isomers of 1 (1 equiv.) with 2 (1.3 equiv.) in the presence of trimethylsilyl triflate (4) (1 equiv.) at -78° C in CH₂Cl₂ afforded after 2H the four diastereomers 5a-5d in ratios shown in Scheme 1. The stereochemistry of these diastereomers were confirmed by comparison with separately prepared authentic specimens (vide infra). The u-isomer gave rise to 93:7 4.5-syn/anti relation whereas the completely opposite outcome resulted with the *l*-counterpart, providing unequivocal evidence for the S_N2-like reaction. In contrast to the reversal in the 4,5-relation, the 3.4-svn preference was observed for both u- and 1-1. One may suppose that the reaction is initiated by allylation of the monothioacetal followed by replacement of the acetoxy group with the resulting organometal thiophenoxide (Scheme 2). The invalidity of this mechanism was proved by the following experiments. The putative intermediate 6 which had been separately prepared was exposed to organotin thiophenoxies, however no reaction occurred at all. The reaction is likely to involve attack of cinnamyltin towards the sp³ carbon of the thioacetal [A] rather than the episulfonium ion [B] as shown in Scheme 1 by analogy to the reaction of allyltins $[9,12^*]$.

The reactions with cinnamylsilane 3 were conducted analogously (Scheme 3). In these cases, both isomers of 1 afforded the 4,5-anti products predominantly. The S_N 1-like mechanism involving an oxocarbenium ion intermediate [C]⁹ is apparent from these results. The reaction with the less nucleophilic cinnamylsilane has a later transition state. Quite naturally, no reaction occurred between 6 and (phenylthio)trimethylsilane (Scheme 2). It should be noted that no isomerization of the starting material prior to the nucleophilic attack has been confirmed in the previous study by checking the stereochemical purities of the recovered starting materials at low conversions [9]. With regard to the 3,4-relation, the syn preference was again observed.

The sense of simple diastereoselection in reactions of acetals with prochiral silyl nucleophiles is usually syn [1,2,4,7] except for some special cases where the anti preference is observed: the reaction between aromatic acetals and crotylsilanes [13], Denmark's intramolecular allylation [14] and Heathcock's reaction employing a bulky acetal and enol silyl ether [15]. The syn selectivity was accounted for in terms of the carbocation intermediacy [4,7,16] the grounds for which are not necessarily explicit, however. The present results sustain that the S_N 1-like mechanism can lead to a considerable level of the syn preference. Notably, the



anti selectivity with bulky acetal and enol silyl ether [15] is likely to emerge from the S_N 1-like mechanism since Heathcock came across this idea on the basis of thionium chemistry [16] and proposed the S_N 1-like mechanism for the reaction of the same acetal with an achiral enol silyl ether [5d]. The reversal in stereochemistry is attributable to a change in the transition state geometry due to steric demands. On the other hand, there appear to be no precedents for stereochemical studies on reactions which unequivocally proceed by an S_N 2-like mechanism. The cinnamyltin reaction described here offers the first such example.

Although it is rather difficult to draw an unambiguous picture of the transition state with the present data alone, we propose, on the basis of steric requirements, a possible explanation as follows. Six transition state geometries are feasible for the 3,4-simple diastereoselection in the S_N2-like mechanism as illustrated in Scheme 4. In the reaction of acetals, the sp^3 carbon to be attacked is concave in striking contrast to the planar sp^2 carbon in carbonyls and, consequently, severe steric interactions arise between the ligands of the acetal carbon and of the nucleophilic carbon of the cinnamyl reagent. Molecular models of 1 tell us that the (acetoxy)phenylmethyl radical is exceedingly larger than methoxyl and hydrogen. Hence, upon nucleophilic attack of cinnamyltin, the largest group of this compound may be aligned so as to occupy the most spacious region that is separated by the methoxyl and hydrogen. Then, let us consider the structure of 2. The phenyl group and olefinic hydrogens are located within the same plane while the allylic hydrogens deviate up and down from the plane. Accordingly, the allylic metal moiety is likely to work as a more sterically demanding group than the phenyl and hydrogen residues when the C=C face approaches the acetal. It follows from these considerations that the geometries D-1 and E-1 are more favored than the others, yet E-1 suffers from steric hindrance between the allylic hydrogens and the methoxyl. In the event, D-1 is postulated to be most favored leading to the preferred 3,4-syn relative stereo-chemistry. By similar treatments, the transition state geometry F may be proposed for the S_N 1-like reaction with cinnamylsilane 3.



2.2. Preparation of authentic samples and stereochemical assignments

Procedures for authentic 5 are shown in Scheme 5. The 4-hydroxy precursor 7 was obtained by allylation of

phenyl(phenylthio)acetaldehyde [17]. A mixture of 3,4syn isomers 7a and 7c was prepared by use of cinnamyltributyltin in the presence of $BF_3 \cdot OEt_2$ [18] while 3,4-anti products were obtained with cinnamyl chloride in the presence SnCl₂-Al [19]. The 4,5-relative stereochemistry of these compounds was determined on the basis of vicinal coupling constants $J_{H4,H5}$; the values for the syn isomers (6.59 and 6.96 Hz for 7a and 7c, respectively) are larger than that for the anti isomers, 7b and 7d (4.76 Hz) [20]. The 3,4-relative stereochemistry was determined by reductive desulfurization of 7 to 8. The stereochemistry of 8 was unambiguously determined by comparison with separately prepared compounds: 8a was obtained by the reaction of phenylacetaldehyde with cinnamyltributyltin in the presence of $BF_3 \cdot OEt_2$ while **8b** was obtained by reaction of phenylacetaldehyde with cinnamyl chloride in the presence of SnCl₂-Al. Finally, O-methylation of 7 afforded 5.

3. Experimental details

3.1. Reaction of (1S*,2S*)-1-acetoxy-2-methoxy-1-phenyl-2-(phenylthio)ethane (u-1) and cinnamyltributyltin

A mixture of u-1 (302 mg, 1 mmol), cinnamyltributyltin (529 mg, 1.3 mmol), TMSOTF (222 mg, 1 mmol) and CH₂Cl₂ (5 ml) was stirred at -78° C for 2H. To this mixture, pyridine (0.3 ml) and aqueous NaHCO₃ were added. The mixture was extracted with ethyl acetate. The organic layer was dried (Na₂SO₄) and concentrated. HPLC analysis showed that the crude product consisted of **5a**, **5b**, **5c** and **5d** in a



Scheme 5.

75.4:17.3:6.3:1.0 ratio. Column chromatography (99:1 hexane/ethyl acetate) gave the mixture of 5 (251 mg, 70%).

3.2. Reaction of (1S*,2R*)-1-acetoxy-2-methoxy-1phenyl-2-(phenylthio)ethane (l-1) and cinnamyltributyltin in the presence of TMSOTf

The reaction of *l*-1 (302 mg, 1 mmol) with cinnamyltributyltin (529 mg, 1.3 mmol) in the presence of TM-SOTf (222 mg, 1 mmol) at -78° C for 2H was carried out in a similar manner as described above. Usual work-up and column chromatography afforded 5 (251 mg, 70%); 5a:5b:5c:5d = 2.6:0.2:73.2:24.0 based on HPLC.

3.3. Reaction of u-1 and cinnamyltrimethylsilane in the presence of TMSOTf

A mixture of u-1 (302 mg, 1 mmol), cinnamyltrimethylsilane (247 mg, 1.3 mmol), TMSOTf (222 mg, 1 mmol) and CH_2Cl_2 (5 ml) was stirred at $-78^{\circ}C$ for 2H and at $-50^{\circ}C$ for 1H and then at $-20^{\circ}C$ for 2H. Usual workup and column chromatography provided 5 (176 mg, 49%); 5a: 5b: 5c: 5d = 27.7:8.4:55.3:8.6.

3.4. Reaction of I-1 and cinnamyltrimethylsilane

The reaction of l-1 (302 mg, 1 mmol) with cinnamyltrimethylsilane (247 mg, 1.3 mmol) in the presence of TMSOTf (222 mg, 1 mmol) was carried out in a similar manner as described above. Usual workup and column chromatography afforded 5 (191 mg, 53%); 5a:5b:5c:5d = 26.5:8.3:57.3:7.9 based on HPLC.

3.5. Synthesis of $(3S^*, 4R^*, 5S^*)$ - and $(3R^*, 4S^*, 5S^*)$ -4hydroxy-3,5-diphenyl-5-phenylthio-1-pentene (7a and 7c)

A mixture of phenyl(phenylthio)acetaldehyde (684 mg, 3 mmol), cinnamyltributyltin (1.58 9, 3.9 mmol), $BF_3 \cdot OEt_2$ (426 mg, 3 mmol), and CH_2Cl_2 (13 ml) was stirred at -78° C for 3 h. To this mixture was added aqueous NaHCO₃ (3 ml) and the mixture was extracted with ethyl acetate. The organic layer was dried (Na_2SO_4) and evaporated to give an oil. Column chromatography (50:50 hexane/benzene) afforded a mixture of 7a and 7c (519 mg, 50%); 7a:7c = 17:83 based on HPLC. Preparative HPLC provided pure 7a and 7c. **7a**: ¹H NMR (CDCl₃): δ 2.54 (d, J = 2.56 Hz, 1H); 3.55 (dd, J = 6.23, 7.70 Hz, 1H); 4.11 (d, J = 6.59 Hz, 1H); 4.24 (ddd, J = 2.56, 6.23, 6.59 Hz, 1H); 5.08 (dd, J =1.47, 17.2 Hz, 1H); 5.11 (dd, J = 1.47, 10.2 Hz, 1H); 6.16 (ddd, J = 7.70, 10.2, 17.2 Hz, 1H); 7.1–7.4 (m, 15H). ¹³C NMR (CDCl₃): δ 53.4, 58.6, 76.5, 116.6, 126.9, 127.34, 127.37, 128.35, 128.39, 128.43, 128.6, 129.2, 132.7, 133.6, 139.1, 139.3, 140.3. MS: (m/z) 346 (M^+) . HRMS: $C_{23}H_{22}OS$ calcd.: (M^+) 346.1391.

Found: 346.1256. IR (CCl₄): 3500, 3600 cm⁻¹. 7c: ¹H NMR (CDCl₃): δ 2.01 (d, J = 3.29 Hz, 1H); 3.43 (t, J = 8.06 Hz, 1H); 4.27 (d, J = 4.76 Hz, 1H); 4.29 (ddd, J = 3.29, 4.76, 8.06 Hz, 1H); 5.01 (dd, J = 1.83, 17.2 Hz, 1H); 5.13 (dd, J = 1.83, 10.2 Hz, 1H); 6.01 (ddd, J = 8.06, 10.2, 17.2 Hz, 1H); 7.2–7.4 (m, 15H). ¹³C NMR (CDCl₃): δ 53.4, 56.7, 74.9, 117.1, 126.8, 127.4, 127.5, 128.2, 128.5, 128.6, 128.8, 129.6, 132.4, 134.0, 137.6, 138.2, 140.2. MS: (m/z) 346 (M⁺). HRMS: C₁₇H₁₆O calcd.: (M⁺ - C₆H₅SH) 236.1201. Found: 236.1024. IR (CCl₄): 3680 cm⁻¹.

3.6. Synthesis of (3R*,4R*,5S*)- and (3S*,4S*,5S*)-4hydroxy-3,5-diphenyl-5-phenylthio-1-pentene (7b and 7d)

To a mixture of SnCl₂ (190 mg, 1 mmol), Al (54 mg, 2 mmol), THF (2 ml) and H_2O (1 ml) were added phenyl(phenylthio)acetaldehyde (228 mg, 1 mmol) and cinnamyl chloride (229 mg, 1.5 mmol) at 40°C. The reaction mixture was stirred at the same temperature for 12H. Extraction with ethyl acetate, drying (Na_2SO_4) the organic layer, and evaporation left an oil. Column chromatography (50:50 hexane/benzene) of the residue afforded a mixture of **7b** and **7d** (221 mg, 64%); 7b:7d = 57:43 based on HPLC. Preparative HPLC gave pure 7b and 7d. 7b: ¹H NMR (CDCl₃): δ 2.78 (d, J = 2.57 Hz, 1H); 3.49 (dd, J = 5.50, 9.16 Hz, 1H); 4.08 (d, J = 6.96 Hz, 1H); 4.16 (ddd, J = 2.57, 5.50, 6.96 Hz, 1H); 5.05 (dd, J = 1.83, 17.2 Hz, 1H); 5.22 (dd, J = 1.83, 9.89 Hz, 1H); 6.23 (ddd, J = 9.16, 9.89, 17.2 Hz, 1H); 7.1-7.4 (m, 15H). ¹³C NMR (CDCl₃): δ 53.1, 58.8, 76.7, 118.1, 126.7, 127.2, 127.3, 127.9, 128.1, 128.4, 128.5, 128.6, 132.5, 133.8, 136.5, 140.3, 141.7. MS: (m/z) 346 (M⁺). HRMS: $C_{17}H_{16}O$ calcd.: (M⁺ - $C_{6}H_{5}SH$) 236.1201. Found: 236.1204. IR (CCl₄): 3500, 3570 cm⁻¹. 7d: ¹H NMR (CDCl₃): δ 2.33 (d, J = 2.94 Hz, 1H); 3.36 (t, J = 8.06 Hz, 1H); 4.09 (d, J = 4.76 Hz, 1H); 4.22 (ddd, J = 2.94, 4.76, 8.06 Hz, 1H); 5.02 (dd, J =1.83, 17.2 Hz, 1H); 5.12 (dd, J = 1.83, 10.2 Hz, 1H); 6.08 (ddd, J = 8.06, 10.2, 17.2 Hz, 1H); 7.0-7.4 (m, 15H). ¹³C NMR (CDCl₃): δ 53.5, 56.1, 74.5, 117.4, 126.8, 127.3, 127.6, 128.2, 128.6, 128.8, 129.6, 132.3, 134.0, 137.3, 138.2, 140.7. MS: (m/z) 346 (M⁺); HRMS: C23H22OS calcd.: (M⁺) 346.1391. Found: 346.1347. IR (CCl_4) : 3350, 3630 cm⁻¹.

3.7. Determination of the C-3 / C-4 relative stereochemistry of 7

3.7.1. Treatment of 7a and 7c with Raney-Nickel

A mixture of 7a (30 mg, 0.087 mmol), Raney-Nickel (W2) (300 mg), and EtOH (3 ml) was refluxed for 5 h. The insoluble material was removed by filtration and the filtrate was evaporated to give an oil. Column chromatography of this oil (95:5 hexane/ethyl ac-

etate) afforded $(3S^*, 4S^*)$ -4-hydroxy-3,5-diphenyl-1pentene (8a) (14 mg, 68%). ¹H and ¹³C NMR spectra were identical with the authentic sample.

Treatment of 7c (50 mg, 0.144 mmol) with Raney-Nickel (W2) (500 mg) in EtOH (3 ml) provided **8a** (21 mg, 77%).

3.7.2. Reaction of 7b and 7d with Raney-Nickel

A mixture of **7b** (45 mg, 0.13 mmol), Raney-Nickel (W2) (500 mg) and EtOH (3 ml) was refluxed for 6 h. Usual workup and column chromatography (95:5 hexane/ethyl acetate) afforded $(3R^*, 4S^*)$ -4-hydroxy-3,5-diphenyl-1-pentene (**8b**) (20 mg, 65%). ¹H and ¹³C NMR spectra were identical with the authentic sample.

Treatment of 7d (30 mg, 0.087 mmol) with Raney-Nickel (400 mg) described above gave 8b (15 mg, 72%).

3.8. Preparation of authentic 8

3.8.1. (3S*,4S*)-4-Hydroxy-3,5-diphenyl-1-pentene (8a)

A mixture of phenylacetaldehyde (120 mg, 1 mmol), cinnamyltributyltin (528 mg, 1.3 mmol), BF₃ · OEt₂ (141 mg, 1 mmol) and CH_2Cl_2 (5 ml) was stirred at $-78^{\circ}C$ for 3 h. Aqueous NaHCO₃ was added to this solution, and the mixture was extracted with ethyl acetate. The organic layer was dried (Na₂SO₄) and concentrated. The crude product proved to consist of $(3S^*, 4S^*)$ - and $(3R^*, 4S^*)$ -4-hydroxy-3,5-diphenyl-1-pentene (8a and 8b, respectively) in an 83:17 ratio based on HPLC. Column chromatography (50:50 hexane/benzene) gave 4-hydroxy-3,5-diphenyl-1-pentene (119 mg, 50%). Preparative HPLC provided pure 8a: ¹H NMR $(CDCl_3)$: δ 1.58 (d, J = 3.66 Hz, 1H); 2.56 (dd, J = 9.53, 13.9 Hz, 1H); 2.97 (dd, J = 2.57, 13.9 Hz, 1H); 3.36 (dd, J = 7.69, 8.79 Hz, 1H); 4.08 (dddd, J = 2.57, 3.66, 7.69,9.53 Hz, 1H); 5.1-5.2 (m, 2H); 6.15 (m, 1H); 7.1-7.4 (m, 10H). ¹³C NMR (CDCl₃): δ 41.0, 56.6, 75.5, 116.9, 126.2, 126.7, 128.3, 128.5, 128.6, 129.3, 138.5, 138.7, 140.6. MS: (m/z) 238 (M⁺). HRMS: C₁₇H₁₈O (M⁺) 238.1350. Found: 238.1503.

3.8.2. $(3R^*, 4S^*)$ -4-hydroxy-3,5-diphenyl-1-pentene (**8b**)

To a suspension of SnCl₂ (379 mg, 2 mmol) and Al (108 mg, 4 mmol) in THF/H₂O (6 ml, 2:1) were added phenylacetaldehyde (240 mg, 2 mmol) and cinnamyl chloride (458 mg, 3 mmol) at 40°C. The reaction mixture was stirred for 5 h. Extraction with ethyl acetate, drying the organic layer (Na₂SO₄), and evaporation left an oil. Column chromatography (50:50 hexane/benzene) of this oil gave 4-hydroxy-3,5-diphenyl-1-pentene (304 mg, 64%); **8b:8a** = 92:8 based on HPLC. Preparative HPLC provided pure **8b**: ¹H NMR (CDCl₃): δ 1.83 (br, 1H); 2.56 (dd, J = 8.79, 13.9 Hz,

1H); 2.72 (dd, J = 3.66, 13.9 Hz, 1H); 3.31 (dd, J = 7.69, 8.79 Hz, 1H); 4.04 (ddd, J = 3.66, 7.69, 8.79 Hz, 1H); 5.18 (dd, J = 1.47, 17.2 Hz, 1H); 5.23 (dd, J = 1.47,10.2 Hz, 1H); 6.18 (ddd, J = 8.79,10.2,17.2 Hz, 1H); 7.1–7.4 (m, 10H). ¹³C NMR (CDCl₃): δ 41.0, 56.4, 75.0, 117.8, 126.2, 126.7, 128.0, 128.3, 128.7, 129.2, 137.9, 138.6, 141.5. MS: (m/z) 238 (M⁺). HRMS: C₁₇H₁₆ calcd.: (M⁺- H₂O) 220.1252. Found: 220.1312.

3.9. Preparation of authentic 4-methoxy-3,5-diphenyl-5phenylthio-1-pentene (5): methylation of 7 with MeI

To a THF suspension (1 ml) of KH (25% mineral oil suspension, 45 mg, 0.28 mmol) was added 7c (50 mg, 0.14 mmol) in THF (0.5 ml) at -20° C. After the mixture had been stirred for 10 min, MeI (40 mg, 0.28 mmol) was added. The mixture was stirred at -20° C for 30 min and quenched with aqueous NH₄Cl. The mixture was extracted with ethyl acetate. The organic layer was dried (Na₂SO₄) and evaporated. Column chromatography (99:1 hexane/ethyl acetate) of the residue afforded $(3R^*, 4S^*, 5S^*)$ -4-methoxy-3,5diphenyl-5-phenylthio-1-pentene (5c) (47 mg, 93%). ¹H NMR (CDCl₃): δ 3.03 (s, 3H); 3.48 (dd, J = 7.69, 8.06 Hz, 1H); 3.83 (dd, J = 5.49, 7.69 Hz, 1H); 4.21 (d, J = 5.49 Hz, 1H); 5.03 (dd, J = 1.46, 17.2 Hz, 1H); 5.12 (dd, J = 1.46, 10.2 Hz, 1H); 6.09 (ddd, J = 8.06, 10.2, 17.2 Hz, 1H); 7.1-7.4 (m, 15H). ¹³C NMR (CDCl₃): δ 53.9, 56.6, 61.7, 87.0, 116.8, 126.4, 126.8, 127.1, 127.8, 128.1, 128.6, 128.8, 129.7, 131.7, 134.9, 138.6, 138.9, 140.8. MS: (m/z) 360 (M⁺). HRMS: C₂₄H₂₄OS calcd.: (M⁺) 360.1548. Found: 360.1545. HPLC (t_R) 18.1 min (Develosil Si 30, 3 μ , 4.6 mm \times 250 mm, 99:1 hexane/ ethyl acetate, 1 ml min $^{-1}$).

Reaction of **7a** with MeI described above provided **5a** in 98% yield. ¹H NMR (CDCl₃): δ 2.99 (s, 3H); 3.72–3.76 (m, 2H); 4.32 (d, J = 3.67 Hz, 1H); 5.12 (dd, J = 1.46,10.2 Hz, 1H); 5.16 (dd, J = 1.46, 17.2 Hz, 1H); 6.18 (m, 1H); 7.0–7.4 (m, 15H). ¹³C NMR (CDCl₃): δ 54.3, 57.4, 61.8, 89.5, 116.9, 126.3, 126.5, 127.1, 128.24, 128.28, 128.4, 128.5, 128.9, 131.2, 135.5, 138.7, 140.7, 141.2. MS: (m/z) 360 (M⁺). HRMS: C₁₈H₁₈O calcd.: (M⁺ - C₆H₅SH) 250.1358. Found: 250.1458. HPLC (t_R) 23.4 min.

Methylation of **7b** gave **5b** in 99% yield. ¹H NMR (CDCl₃): δ 3.23 (s, 3H); 3.62 (dd, J = 6.23, 8.79 Hz, 1H); 3.74 (dd, J = 5.86, 6.23 Hz, 1H); 4.14 (d, J = 5.86Hz, 1H); 5.00 (dd, J = 1.83, 16.8 Hz, 1H); 5.18 (dd, J = 1.83, 9.89 Hz, 1H); 6.28 (ddd, J = 8.79, 9.89, 16.8 Hz, 1H); 7.05-7.37 (m, 15H). ¹³C NMR (CDCl₃): δ 53.5, 57.7, 62.1, 89.7, 117.7, 126.3, 126.6, 127.1, 128.1, 128.33, 128.35, 128.4, 128.6, 131.3, 135.3, 136.9, 140.8, 142.0. MS: (m/z) 360 (M⁺). HRMS: C₁₈H₁₈O calcd.: (M⁺-C₆H₅SH) 250.1358. Found: 250.1377. HPLC ($t_{\rm R}$) 28.4 min. Treatment of 7d with MeI provided 5d in 95% yield. ¹H NMR (CDCl₃): δ 3.21 (s, 3H); 3.55 (dd, J = 6.59, 7.33 Hz, 1H); 3.76 (dd, J = 5.50, 6.59 Hz, 1H); 4.10 (d, J = 5.50 Hz, 1H); 4.98 (dd, J = 1.83, 16.8 Hz, 1H); 5.11 (dd, J = 1.83, 9.89 Hz, 1H); 6.19 (ddd, J = 7.33, 9.89, 16.8 Hz, 1H); 7.1–7.4 (m, 15H). ¹³C NMR (CDCl₃): δ 53.5, 56.1, 61.9, 86.9, 117.2, 126.6, 126.8, 127.2, 127.8, 128.3, 128.5, 128.7, 129.6, 131.7, 134.9, 137.6, 138.8, 141.6. MS: (m/z) 360 (M⁺). HRMS: C₂₄H₂₄OS calcd.: (M⁺) 360.1548. Found: 360.1459. HPLC (t_R) 24.8 min.

3.10. Preparation of $(3R^*, 4S^*, 5S^*)$ -5-acetoxy-4-methoxy-3,5-diphenyl-1-pentene

A mixture of methyl mandelate (16.6 g, 100 mmol), t-butyldimethylsilyl chloride (22.6 g, 150 mmol), imidazole (20.4 g, 300 mmol) and DMF (200 ml) was stirred at room temperature for 29 h. Extractive workup and column chromatography on silica gel (97:3 hexane/ ethyl acetate) provided methyl (R^*)-(t-butyldimethylsiloxy)phenylethanoate (18.1 g, 65 %). ¹H NMR (CDCl₃): δ 0.01 (s, 3H); 0.09 (s, 3H); 0.90 (s, 9H); 3.66 (s, 3H); 5.22 (s, 1H); 7.2–7.5 (m, 5H). ¹³C NMR (CDCl₃): δ –5.2, –5.1, 18.2, 25.6, 52.0, 74.3, 126.2, 128.0, 128.2, 139.0, 172.5. MS: (m/z) 280 (M⁺). HRMS: C₁₅H₂₃O₃Si calcd.: (M⁺ – H) 279.1417. Found: 279.1497.

To a solution of methyl (R^*)-(t-butyldimethylsiloxy)phenylethanoate (8.4 g, 30 mmol) in toluene (60 ml) was added diisobutylaluminum hydride (1.0 M hexane solution, 33 ml, 33 mmol) at -78° C. After stirring at the same temperature for 1 h, the mixture was quenched with saturated aqueous NH₄Cl (5 ml). Extractive workup with ether afforded (S^*)-(t-butyldimethylsiloxy)phenylethanal (8.03 g) which was used in the next step without purification. ¹H NMR (CDCl₃): δ 0.04 (s, 3H); 0.11 (s, 3H); 0.94 (s, 9H); 5.00 (d, J = 2.19 Hz, 1H); 7.2–7.5 (m, 5H); 9.51 (d, J = 2.19 Hz, 1H). ¹³C NMR (CDCl₃): δ –4.8, 18.3, 25.7, 79.9, 126.3, 128.3, 128.7, 139.1, 199.4. MS: (m/z) 250 (M⁺). HRMS: C₁₃H₁₉O₂Si calcd.: (M⁺ – CH₃) 235.1154. Found: 235.1252.

A dichloromethane solution of $BF_3 \cdot OEt_2$ (2 M solution, 15 ml, 30 mmol) was added to a mixture of (S^*) -(t-butyldimethylsiloxy)phenylethanal (7.5 g, 30 mmol) described above, cinnamyltributyltin (16.3 g, 40 mmol), and CH_2Cl_2 (100 ml) at $-78^{\circ}C$. The mixture was stirred at $-78^{\circ}C$ for 2 h and then worked up to give an oil. Column chromatography of this oil on silica gel (97:3 hexane/ethyl acetate) gave ($3R^*, 4S^*, 5S^*$)-5-t-butyldimethylsiloxy-4-hydroxy-3,5-diphenyl-1-pen-

tene (4.82 g, 44%). The 3,4,5-relative stereochemistry of this compound was determined as follows. (1) The C_3 - C_4 stereochemistry was tentatively assigned to be

syn on the basis of the reaction mechanism. (2) A small portion of this compound was treated with Bu₄NF (3.3 equiv.) in THF at room temperature for 5 h to afford $(3R^*, 4S^*, 5S^*)$ -4,5-dihydroxy-3,5-diphenyl-1-pentene in 87% yield. ¹H NMR (CDCl₃): δ 2.17 (br, 1H); 2.74 (br, 1H); 3.47 (t-like, J = 7.69 Hz, 1H); 3.99–4.08 (m, 1H); 4.64 (d, J = 3.97 Hz, 1H); 5.09–5.17 (m, 2H); 6.01–6.14 (m, 1H); 7.2–7.4 (m, 10H). ¹³C NMR (CDCl₃): δ 52.7, 73.5, 77.5, 117.0, 126.5, 126.9, 127.7, 128.4, 128.6, 128.8, 138.6, 140.0, 141.6. The 4,5-anti stereochemistry was determined on the basis of $J_{H4,H5} = 3.97$ Hz [21]. IR (CCl₄): 3576 cm⁻¹. 1H NMR (CDCl₃): δ 0.06 (s, 3H); 0.13 (s, 3H); 0.90 (s, 9H); 2.90 (d, J = 4.27 Hz, 1H); 3.26 (t-like, J = 7.15 Hz, 1H); 3.94–4.01 (m, 1H); 4.54 (d, J = 5.62 Hz, 1H); 4.97–5.15 (m, 2H); 6.06–6.20 (m, 1H); 7.2–7.7 (m, 10H). ¹³C NMR (CDCl₃): δ –5.1, -4.4, 18.0, 25.7, 52.0, 74.3, 75.6, 78.6, 115.7, 126.2, 126.4, 127.1, 127.7, 128.0, 128.1, 128.2, 129.2, 139.0, 139.9, 140.9, 141.6. MS: (m/z) 368 (M⁺).

To a suspension of KH (35% oil dispersion, 1.26 g, 11 mmol) in THF (30 ml) was added $(3R^*, 4S^*, 5S^*)$ -5t-butyldimethylsiloxy-4-hydroxy-3,5-diphenyl-1-pentene (2.58 g, 7 mmol) in THF (1 ml) at 0°C. After 10 min, MeI (0.92 ml, 14 mmol) was added. The resulting mixture was stirred at room temperature for 17 h and then worked up. Column chromatography on silica gel (99:1 hexane/ethyl acetate) provided $(3R^*, 4S^*, 4S^*)$ 5S*)4-t-butyldimethylsiloxy-5-methoxy-3,5-diphenyl-1pentene (47 mg, 2%) and (3R*,4S*,5S*)-5-t-butyldimethylsiloxy-4-methoxy-3,5-diphenyl-1-pentene (1.14 g, 43%). $(3R^*, 4S^*, 5S^*)$ -4-t-butyldimethylsiloxy-5-methoxy-3,5-diphenyl-1-penten e. ¹H NMR (CDCl₃): δ 0.00 (s, 3H); 0.01 (s, 3H); 0.94 (s, 9H); 2.9-3.05 (m, 1H); 3.02 (s, 3H); 3.79 (d, J = 7.45 Hz, 1H); 4.17 (dd, J =3.66, 7.45 Hz, 1H); 4.78 (d, J = 17.2 Hz, 1H); 4.97 (d, J = 9.89 Hz, 1H); 6.09–6.13 (m, 1H); 7.0–7.4 (m, 10H). ¹³C NMR (CDCl₃): δ -4.8, -4.1, 18.6, 26.3, 52.7, 56.0, 80.3, 86.1,115.0,126.3,127.8,128.1, 128.3, 130.1, 139.1, 140.5, 141.5. MS: (m/z) 382 (M⁺). HRMS: $C_{23}H_{31}O_2Si$ calcd.: $(M^+ - CH_3)$ 367.2097. Found: 367.2015. ($3R^*, 4S^*, 5S^*$)-5-t-butyldimethylsiloxy-4methoxy-3,5-diphenyl-1-penten e. ¹H NMR (CDCl₃): δ 0.17 (s, 3H); 0.26 (s, 3H); 1.09 (s, 9H); 3.39-3.45 (m, 1H); 3.49 (s, 3H); 3.74 (t-like, J = 6.04 Hz, 1H); 4.83 (d, J = 6.04 Hz, 1H); 5.13 (d, J = 17.1 Hz, 1H); 5.26 (d, J = 11.5 Hz, 1H); 6.30–6.44 (m, 1H); 7.4–7.7 (m, 10H). ¹³C NMR (CDCl₃): δ -5.1, -4.4, 18.0, 25.7, 52.0, 74.3, 75.6, 78.7, 115.8, 126.2, 126.4, 127.1, 127.8, 128.1, 128.7, 129.3, 139.9, 140.6, 141.7. MS: (m/z) 382 (M⁺). HRMS: C₂₃H₃₁O₂Si (M⁺ - CH₃O) 351.2144. Found: 351.2220. A mixture of $(3R^*, 4S^*, 5S^*)$ -5-t-butyldimethylsiloxy-4-methoxy-3,5-diphenyl-1-pentene (1.15 g, 3 mmol), Bu₄NF (1 M THF solution, 6 ml, 6 mmol) and THF (15 ml) was stirred at room temperature for 38 h. Extractive workup with ethyl acetate and column chromatography on silica gel (90:10 benzene/ethyl acetate) gave ($3R^*, 4S^*, 5S^*$)-5-hydroxy-4-methoxy-3,5diphenyl-1-pentene (430 mg, 53%). IR (CCl₄): 3572 cm^{-1. 1}H NMR (CDCl₃): δ 2.82 (br, 1H); 3.02 (s, 3H); 3.45-3.61 (m, 2H); 4.69 (d, J = 3.42 Hz, 1H); 5.08-5.18 (m, 2H); 6.10-6.24 (m, 1H); 7.2-7.4 (m, 10H). ¹³C NMR (CDCl₃): δ 52.4, 61.2, 73.1, 88.9, 116.7, 126.2, 126.3, 127.3, 128.1, 128.7, 138.7, 140.8, 142.2. MS: (m/z) 268 (M⁺). HRMS: C₁₈H₂₀O₂ calcd.: (M⁺) 268.1463. Found: 268.1486.

Reaction of $(3R^*, 4S^*, 5S^*)$ -5-hydroxy-4-methoxy-3,5-diphenyl-1-pentene (379 mg, 1.5 mmol) with acetic anhydride (1.0 ml, 21.6 mmol) and 4-(*N*,*N*-dimethylamino)pyridine (20 mg) in pyridine (3 ml) at room temperature for 4 h provided **6** (353 mg, 80%). IR (CCl₄): 1746 cm^{-1.} ¹H NMR (CDCl₃): δ 2.11 (s, 3H); 2.90 (s, 3H); 3.41 (t-like, J = 7.73 Hz, 3H); 3.65 (dd, J = 4.33 Hz, 1H); 6.04-6.20 (m, 1H); 7.2-7.4 (m, 10H). ¹³C NMR (CDCl₃): δ 20.9, 52.8, 61.5, 76.1, 87.4, 116.5, 126.8, 128.1, 128.2, 128.6, 138.1, 140.5, 169.6. MS: (*m/z*) 310 (M⁺). HRMS: C₁₈H₁₇O calcd.: (M⁺-MeCOOH – H) 249.1279. Found: 249.1186.

3.11. Reaction of 6 with $PhSSiMe_3$ or $PhSSnBu_3$ in the presence of TMSOTf

3.11.1. With PhSSiMe₃

To a solution of 6 (29.4 mg, 0.1 mmol), PhSSiMe₃ (0.5 M CH₂Cl₂ solution, 0.3 mi, 0.15 mmol) in CH₂Cl₂ (1 ml) was added TMSOTf (1 M CH₂Cl₂ solution, 0.2 ml, 0.2 mmol) at -78° C. The resulting mixture was stirred at -78° C for 1 h, -50° C for 1 h, and -20° C for 1 h and then quenched with saturated aqueous NaHCO₃ (1 ml). Extractive workup provided an oil (43 mg). TLC, HPLC and ¹H NMR analyses of this oil indicated that no 4-methoxy-3,5-diphenyl-5-phenylthio-1-pentene was formed. Column chromatography on silica gel (90:10 benzene/ethyl acetate) afforded 6 (26 mg) which was identical with an authentic sample.

3.11.2. With PhSSnBu₃

Treatment of 6 (29.4 mg, 0.1 mmol) with Bu₃SnSPh (0.5 M CH₂Cl₂ solution, 0.3 ml, 0.15 mmol) and TM-SOTF (1 M CH₂Cl₂ solution, 0.2 ml, 0.2 mmol) in CH₂Cl₂ (1 ml) at -78° C for 1 h, -50° C for 1 h, and -20° C for 1 h afforded an oil (78 mg) which was chromatographed on silica gel (90:10 benzene/ethyl

acetate) to give 6 (24 mg). No 4-methoxy-3,5-diphenyl-5-phenylthio-1-pentene was produced.

References

- T. Mukaiyama and M. Murakami, Synthesis, (1987) 1043, I. Fleming, J. Dunogues and R. Smithers, Org. React., 37 (1989) 57; H. Sakurai, Pure Appl. Chem., 54 (1982) 1; A. Alexakis and P. Mangeney, Tetrahedron: Asymmetry, 1 (1990) 477; Y. Yamamoto and N. Asao, Chem. Rev., 93 (1993) 2207.
- 2 (a) S.E. Denmarak and N.G. Almstead, J. Am. Chem. Soc., 113 (1991) 8089 and refs. therein; (b) T. Sammakia and R.S. Smith, J. Org. Chem., 57 (1992) 2997.
- 3 S.E. Denmark, T.M. Willson, N.G. Almstead, J. Am. Chem. Soc., 111 (1989) 9258; S.E. Denmark and T.M. Willson, in D. Schinzer (ed.), Selectivities in Lewis Acid Promoted Reactions, Kluwer, Norwell, 1989, p. 247.
- 4 S. Murata, M. Suzuki and R. Noyori, Tetrahedron, 44 (1988) 4259.
- 5 For further leading references: (a) P.A. Bartlett, W.S. Johnson and J.D. Elliott, J. Am. Chem. Soc., 105 (1983) 2088; (b) T. Harada, T. Hayashiya, I. Wada, N. Iwa-ake and A. Oku, J. Am. Chem. Soc., 109 (1987) 527; (c) K. Ishihara, A. Mori and H. Yamamoto, Tetrahedron, 46 (1990) 4595; (d) I. Mori, K. Ishihara, L.A. Flippin, K. Nozaki, H. Yamamoto, P.A. Bartlett and C.H. Heathcock, J. Org. Chem., 55 (1990) 6107.
- 6 For an MO study, see: J.L. Broeker, R.W. Hoffmann and K.N. Houk, J. Am. Chem. Soc., 113 (1991) 5006.
- 7 For recent relevant studies, see: J.S. Panek and M.J. Yang, J. Am. Chem. Soc., 113 (1991) 6594; J. Org. Chem., 56 (1991) 5755.
- 8 Acetal: Y. Yamamoto, H. Yatagai, Y. Ishihara, N. Maeda and K. Maruyama, *Tetrahedron, 40* (1984) 2239. Dithioacetal: B.M. Trost and T. Sato, J. Am. Chem. Soc., 107 (1985) 719. Monothioacetal: T. Sato, S. Okura, J. Otera and H. Nozaki, *Tetrahedron Lett., 28* (1987) 6299.
- 9 T. Sato, J. Otera and H. Nozaki, J. Org. Chem., 55 (1990) 6116.
- 10 S.E. Denmark, E.J. Weber, T. Willson and T.M. Willson, *Tetra*hedron, 45 (1989) 1053.
- 11 No isomerization of the cinnamyl compounds was confirmed under the reaction conditions employed in this study.
- 12 Allyltrimethylsilane furnished the 4,5-syn products [9]. The 4,5-relation will be discussed in a separate paper.
- 13 A. Hosomi, M. Ando and H. Sakurai, Chem. Lett., (1986) 365.
- 14 S.E. Denmark and T.M. Willson, J. Am. Chem. Soc., 111 (1989) 3475.
- 15 I. Mori, K. Ishihara and C.H. Heathcock, J. Org. Chem., 55 (1990) 1114.
- 16 For thionium intermediates, see: I. Mori, P.A. Bartlett and C.H. Heathcock, J. Org. Chem., 55 (1990) 5966.
- 17 T. Sato, H. Okazaki, J. Otera and H. Nozaki, J. Am. Chem. Soc., 110 (1988) 5209.
- 18 Y. Yamamoto, H. Yatagai, Y. Naruta and K. Maruyama, J. Am. Chem. Soc., 102 (1980) 7107; Y. Yamamoto, Aldrichim. Acta, 20 (1987) 45; W.R. Roush, in B.M. Trost (ed.), Comprehensive Organic Synthesis, Vol. 2, Pergamon, Oxford, 1991, p. 1.
- 19 J.M. Coxon, S.J. van Eyk and P.J. Steel, Tetrahedron Lett., 6121 (1985) 6121; Tetrahedron, 45 (1989) 1029.
- 20 M. Shimagaki, T. Meada, Y. Matsuzaki, I. Hori, T. Nakata and T. Oishi, *Tetrahedron Lett.*, 25 (1984) 4775.
- 21 M. Fujita and T. Hiyama, J. Am. Chem. Soc., 106 (1984) 4629.